# Rational Prioritization Strategy Allows the Design of Macrolide Derivatives that Overcome Antibiotic Resistance

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This manuscript was compiled on September 30, 2021

Antibiotic resistance is a major threat to global health, which can be addressed by the development of new antibacterial agents to keep pace with the evolutionary adaptation of pathogens. Computational approaches are essential tools to this end, since their application enables fast and early strategical decisions in the drug development process. We present a rational design approach, in which novel acylide antibiotics were screened based on computational predictions of solubility, membrane permeability, and binding affinity towards the ribosome. To assess our design strategy, we tested all candidates for in vitro inhibitory activity and then evaluated them in vivo with several antibiotic-resistant strains to determine minimal inhibitory concentrations. The predicted best candidate is synthetically more accessible, exhibits higher solubility and binding affinity to the ribosome, and is up to 56 times more active against resistant pathogens than telithromycin. Notably, the best compounds designed by us show activity, especially when combined with the membrane-weakening drug colistin, against A. baumanii, P. aeruginosa and E. coli, which are the three most critical targets from the priority list of pathogens of the World Health Organization.

acylides | ribosome | free energy calculations |  $\mathrm{QM}/\mathrm{MM}$  | drug resistance

#### 1. Introduction

The resistance of bacterial pathogens against antibiotics is a global problem that requires the constant development of new antibacterial agents.(1, 2) The design of new derivatives, based on the modification of existing antibiotics or natural products, relies on extensive structural optimization.(3) This process can be time-consuming and costly, given the possible synthetic routes, and the need to consider key factors such as binding affinity, pharmacokinetics, and toxicity. Efficient computational strategies are thus essential to address this problem.

Macrolides are an important class of antibiotics that act against the bacterial ribosome. The first described natural macrolide, erythromycin, is the precursor of semi-synthetic modern antibiotics such as azithromycin, clarithromycin, telithromycin and solithromycin. The design of new macrolide antibiotics commonly relies on semi-synthesis and, more recently, on total synthesis. (4–11) The synthetic routes comprise six linear steps in the case of clarithromycin and ten or more linear steps for telithromycin or for fully synthetic macrolides. Macrolides bind a conserved position in the bacterial ribosome and only show a small number of polar contacts with ribosomal RNA.(12, 13) Crystal structures of bound clarithromycin(14) revealed that the cladinose moiety does not directly interact with the ribosome. Thus, modifications of this side chain can be used to address resistance mechanisms. (15–17) As an additional benefit, the cladinose moiety can be replaced by a more

synthesizable aryl acetic acid derivative to form acylides. (5, 6)

Here, we outline a rational design strategy and apply it to find candidates with improved antibacterial activity based on a prioritization scheme that draws inspiration from branch-and-bound optimization algorithms. (18) Candidates are eliminated or prioritized based on theoretical lower and upper bounds for chemical properties that are essential for activity. The protocol provides design decisions as fast and early as possible, aiming for a computational prioritization within a matter of days to guide the synthesis and testing efforts.

Six conditions must be met by a candidate to proceed through the different stages of the prioritization scheme. The molecule must be synthetically accessible (stage I), soluble in aqueous solution (stage IIa), and exhibit sufficient membrane permeability (stage IIb) as well as high binding affinity towards the target (stage IIIa-d). Moreover, candidates should avoid resistance mechanisms (stage IV) and must be nontoxic (stage V). The design process is simplified when the optimization starts from an approved drug, since therapeutic agents generally fulfill most of these conditions. Thus, the focus lies on eliminating modifications that lead to a loss of essential properties, and on favoring candidates for which these prop-

## **Significance Statement**

Due to the development of resistance against commonly used antibiotics, new derivatives that avoid resistance mechanisms need to be developed. To address this problem, a rational prioritization strategy is outlined for macrolide antibiotics. Candidates are screened based on their solubility, membrane permeability, and binding affinity, using a tiered optimization approach of free energy simulations and quantum mechanical calculations. After prioritization by computational methods, the best candidates are evaluated experimentally. The strategy creates a targeted substance library that is highly enriched in compounds with antibacterial activity. This allows faster iterations in the development of new antibiotic derivatives.

G.K. performed the free energy calculations, P.S. conducted the MD simulations and QM/MM calculations, W. T. contributed QM/MM and FEP expertise and supported the work of P.S., G.K. and E.S.G. N.P. contributed the synthesis and spectroscopic analysis of the compounds, D. Möller did the synthesis of the compounds, P.P. and J.B. contributed the MIC tests, S.H. analyzed the experimental data, G.C. and D. Matzov contributed the *in vitro* inhibition experiments. F.S., A.B., A.Y. and E.S.G. designed the project, analyzed the data and supervised the research. P.S., and G.K. wrote the paper with the contributions of D. Möller, S.H and A.B.

The authors declare that they have no conflict of interest.

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erties are improved or at least conserved. From one stage to the next, fewer molecules are evaluated, but with increasing rigor. The rational design starts with a strategy for synthesis, followed by prioritization through a series of computational techniques. The prioritized molecules are then tested both *in vitro* and *in vivo*. The toxicity of the most active candidates is evaluated in the final stage.

**Fig. 1.** Synthesis of the acylide library. CTY denotes clarithomycin. i) 6.1 eq.  $ArCH_2CO_2H$ , 6.0 eq.  $EDC \cdot HCl$ , 1.1 eq. DMAP, DCM, RT or 3.3 eq.  $ArCH_2CO_2H$ , 3.3 eq. PivCl, 3.3 eq.  $El_3N$ , 1 eq. DMAP, DCM,  $-15^{\circ}C \rightarrow RT$ ; ii) MeOH, RT, 48 h; iii) 2 eq.  $NiCl_2 \cdot 6H_2O$ , 4 eq.  $NaBH_4$ , MeOH,  $0^{\circ}C$ . EDC: 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide, DMAP: 4-dimethylaminopyridine, PivCl: trimethylacetyl chloride. Previously described compounds are indicated with \*. Computed structures are highlighted in red.

In this work, we validate our design strategy with the study of 19 acylide replacements of the cladinose moiety of clarithromycin, using four linear synthesis steps (Fig. 1). This is two steps shorter than that for telithromycin. The numbering of the library starts with the original clarithromycin molecule (1, CYT). Its derivatives include substituents with different polarities and electron densities, aliphatic analogs, or inflexible spacers between the macrolide core and the aryl substituent. The set was designed to evaluate the effects of nitrosubstituents, (5, 6) an extended  $\pi$ -system, an electron-donating substituent, a cyclohexyl-derivative (as a non-aromatic comparison), as well as halogenated and methylated derivatives. Both the spatial and electronic constraints on the scaffold are explored.

## Results

All computational results are derived from a consistent set of relative free energy perturbation (FEP) simulations. The ligands were simulated both in the unbound state in aqueous solution, and in the bound state within the bacterial ribosome. The bound states of clarithromycin and the acylide derivatives were modeled with the structure of the 50S ribosome of Deinococcus radiodurans together with explicit water and with an ionic concentration of 0.1 M. The clarithromycin-ribosome complex was equilibrated with molecular dynamics simulations (SI Section A and Fig. S1). In parallel, relative FEP protocols for all ligands were set up and tested in aqueous solution. The

simulations decouple the side chains of the acylides as shown in Fig. 1, leaving only the common backbone as the alchemical intermediate state. This also allows the use of enhanced sampling methods for exploring the conformational flexibility of the side chains.(19)

First, we estimated the relative affinities for water, in the absence of the ribosome, using implicit solvent calculations (stage IIa, SI Section C).(20) These calculations represent an upper bound for their relative solubilities and neglect self-solvation effects,(21) as well as the free energy of sublimation.(22) Stage IIa was motivated by the relatively poor solubility of clarithromycin and the assumption that hydrophilic molecules might be less affected by promiscuous ATP-dependent efflux pumps.(17, 23) Since all molecules were predicted to be more hydrophilic than clarithromycin (SI Table S1), we used the five compounds with the highest affinity for water (6, 3, 8, 5, 4) for the next stage (III, where we calculated the binding free energy of the ligands towards the ribosome).

Likewise, we estimated membrane permeabilities using an implicit membrane model (stage IIb, SI Section C).(24) The resulting transfer free energies gauge membrane permeability based on Fick's First Law,(25) but neglect self-solvation effects,(26) as well as the nanostructure of the membrane. Since all candidates were considered to be membrane-permeable (SI Table S2), the five molecules with the highest affinity for the cell membrane (7, 2, 9, 18, 14) were also prioritized for the binding free energy calculations in stage III. Step IIb was motivated by the assumption that membrane-permeable molecules are more likely to enter bacterial cells.(17)

Accordingly, we selected molecules 2-9, 14, and 18 for FEP binding free calculations (Stage III). The final frame from the molecular dynamics simulations of 1 bound to the ribosome served as the starting point for relative FEP calculations of the bound state (SI Section D). To reduce computational costs, the ribosome was truncated to a subsystem comprising 0.1 million atoms and all atoms outside of the binding pocket were kept frozen. Thus, any slow structural transitions of the ribosome are neglected.

The equilibration phase of the FEP binding simulations was employed to filter out non-binders by post-processing the trajectories of compounds 2-9, 14, and 18 with the 1-trajectory MM-GBSA method (stage IIIa, SI Section D).(27, 28) These types of calculations only provide very approximate estimations of the free energy changes since the structural relaxation of the unbound state is not accounted for and due to the implicit treatment of the solvent. In addition, any slow relaxation of the binding pocket is neglected due to the short simulation length. Nevertheless, this step allowed us to eliminate compounds 14 and 18 because they exhibited highly unfavorable relative binding free energies (SI Table S3).

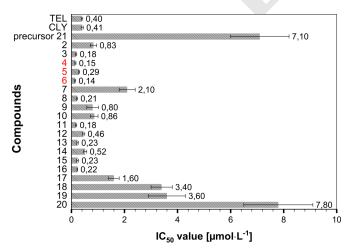
Next, we calculated the relative free energies of binding  $(\Delta\Delta G_{bind})$  and relative free energies of hydration  $(\Delta\Delta G_{hydr})$  of the remaining prioritized set of compounds (2-9) with respect to clarithromycin using a rigorous free energy estimator and explicit solvent (stage IIIb, Table 1, SI Section B). Based on similar FEP protocols in the blind SAMPL prediction challenges, (29, 30) the expected error of these calculations is between 2-3 kcal mol<sup>-1</sup>. Thus, molecule 6 is a significantly better binder than clarithromycin, while molecules 3, 4, 5, and possibly 9 are likely to exhibit improved binding affinities towards the ribosome. The explicit solvent  $\Delta\Delta G_{hydr}$  values

Table 1. Computed relative binding free energies ( $\Delta\Delta G_{bind}$ ), relative hydration free energies ( $\Delta\Delta G_{hydr}$ ), as well as average QM/MM interaction energies in the ribosome ( $\Delta\Delta E_{ribo}$ ) and in aqueous solution ( $\Delta\Delta E_{aq}$ ) for the prioritized acylides with respect to clarithromycin (in kcal mol $^{-1}$ ). The molecules are ordered based on their QM/MM affinity towards the ribosome. The three best candidates are highlighted in red.

Ligand	$\Delta\Delta G_{bind}$	$\Delta\Delta G_{hydr}$	$\Delta \Delta E_{ribo}$	$\Delta \Delta E_{aq}$
6	$-4.3\pm1.5$	$\textbf{-8.5} \pm \textbf{0.9}$	$\textbf{-38.8} \pm \textbf{0.3}$	$\textbf{-15.3} \pm 0.6$
5	$-2.0\pm1.5$	$\textbf{-4.9} \pm \textbf{0.9}$	$\textbf{-34.6} \pm \textbf{0.3}$	$\text{-23.5} \pm 0.6$
4	-2.5 $\pm$ 1.5	$-1.9\pm1.0$	$\textbf{-27.5} \pm \textbf{0.3}$	$\textbf{-4.3} \pm \textbf{0.6}$
3	$-2.4\pm1.5$	$-4.2\pm1.0$	$\textbf{-11.5} \pm 0.3$	$-22.0\pm0.6$
2	$\textbf{-0.3} \pm \textbf{1.5}$	$\textbf{-3.0} \pm \textbf{0.9}$	$\textbf{-6.4} \pm \textbf{0.3}$	$\textbf{3.9} \pm \textbf{0.6}$
9	$-1.9\pm1.5$	$-2.7\pm0.9$	$-2.7\pm0.3$	$-3.2\pm0.6$
8	$\textbf{5.8} \pm \textbf{1.5}$	$-3.1\pm1.0$	$0.8 \pm 0.3$	$\textbf{-8.7} \pm \textbf{0.6}$
7	$12.2 \pm 1.6$	$-3.4\pm0.9$	$11.3 \pm 0.3$	$5.4 \pm 0.6$

confirm that all tested compounds are more hydrophilic than clarithromycin, especially  $\bf 6$ .

In stage IIIc, we carried out QM/MM calculations to further explore the relative stability of the ribosome-ligand complexes in explicit solvent (SI Section E). The relative average interaction energies of the ligands with their surroundings were determined in the ribosome ( $\Delta \Delta E_{ribo}$ ) and in aqueous solution ( $\Delta \Delta E_{aq}$ ). The use of QM/MM approaches previously improved the accuracy of computational free energy predictions by about one kcal mol<sup>-1</sup>.(30, 31) The QM/MM calculations indicate that compounds 6, 5, and 4 exhibit the the most favorable interaction energies with the ribosome. They were, therefore, prioritized for the *in vitro* testing (stage IIId). Molecules 7 and 8 were eliminated because of the predicted instability of their complexes with the ribosome (Table 1). The results for the other molecules are less clear, as stronger interactions with the ribosome are often offset by a simultaneous high affinity for the unbound state in water. Overall, the QM/MM interaction energies confirm the FEP results, indicating that 6 is a very favorable binder to the ribosome. In addition, ligands 5 and 4 emerge as potential effective inhibitors. A detailed discussion of the simulations of the ligand-ribosome complexes is provided in the SI Section E.



**Fig. 2.**  ${\rm IC}_{50}$  data for the inhibition of the *E. coli* ribosome by telithromycin (TEL), clarithromycin (CTY), the precursor molecule (21) and all 19 candidates. The three best candidates from the computational preselection are highlighted in red.

To evaluate our rational design strategy, the inhibitory

activities of all 19 candidates were determined by  $IC_{50}$  values from an *in vitro* transcription-translation assay against the  $E.\ coli$  ribosome (stage IIId, experimental SI, Section  $IC_{50}$  determination). As evidenced in Fig. 2, compounds **6** and **4** are indeed the most active in the whole set, and **5** is more active than clarithromycin. Molecule **7**, which was predicted to be the worst binder in the prioritized set, also turns out the be the least active in this subset. Ligands **11**, **13**, **15** and **16** also show high affinity towards the ribosome, but were not part of the prioritized set because of their solvation properties. Membrane permeability and hydrophobicity are expected to play a prominent role for biological activity.

In stage IV, we tested the acylides for in vitro antimicrobial activity by assessing minimal inhibitory concentrations (MIC) against Gram-positive bacteria (B. subtilis and S. aureus) as well as against Gram-negative bacteria (E. coli, A. baumanii, P. aeruginosa, and K. pneumoniae) and yeast strains (C. albicans and C. neoformans). The MIC results are listed in Fig. 3. We also tested the cytotoxicity and hemolytic activity by assessing  $CC_{50}$  against human kidney cell-line HEK293, and  $HC_{10}$  against human red blood cells (stage V, last two rows of Fig. 3). For MIC values up to  $32\,\mu\mathrm{g}\,\mathrm{mL}^{-1}$  therapeutic indices are calculated ( $CC_{50}$  / MIC) and shown in parentheses below the respective MIC value in Fig. 3. As the ester bond in acylides might be prone to enzymatic breakdown, all acylides were tested to be stable in human blood serum.(33)

All compounds are highly active against the MSSA strain of S. aureus (DSM20231, ATCC 12600)(32) with MIC values up to  $0.1\,\mu\mathrm{g\,mL^{-1}}$ , while displaying no cytotoxicity or hemolytic activity at the same concentrations, as well as no activity against either of the yeast strains. Exceptions are only compounds 19 and 20, the two benzoic acid variants, which have no antibacterial activity. The high antibiotic activity of the other compounds is retained against B. subtilis (DSM402, NCIB 10106), and mostly against two mutant strains of MSSA, S. aureus BAA-976 and BA-977. S. aureus BAA 977 features inducible macrolide-lincosamide-streptogramin B resistance (iMLSB) due to the erythromycin ribosomal methylase erm(C)gene that alters the ribosomal structure, (34) while S. aureus BAA 976 is macrolide-resistant due to the ATP-dependent efflux pump associated with the msr(A) gene. Both strains display resistance against the reference antibiotics erythromycin (ERY), clarithromycin (CTY), and telithromycin (TEL). Most compounds retain similar activity in the resistant strains compared to the MSSA strains. The only exceptions are compound 5, which displays a 10-fold reduction of activity (MIC value shift from 0.1 to  $1 \mu \text{g mL}^{-1}$ ), compound **6** with a 5-fold reduction (MIC value shift from 0.1 to  $0.5 \,\mu\mathrm{g\,mL}^{-1}$ ) and compound 8 with a 4-fold reduction (MIC value shift from 4 to  $16 \,\mu\mathrm{g}\,\mathrm{mL}^{-1}$ ). Compounds 3, 6, and 16 still retain a high antibiotic activity with MIC values  $< 1 \,\mu\mathrm{g}\,\mathrm{mL}^{-1}$  in the resistant strains.

None of the compounds show high activity against the MRSA strain S. aureus ATCC 43300, which harbours, among other resistant genes, the erm(A) gene. The only notable exception is compound  $\mathbf{7}$ , displaying some low activity with a MIC value of  $32 \,\mu\mathrm{g}\,\mathrm{mL}^{-1}$ , compared to  $8 \,\mu\mathrm{g}\,\mathrm{mL}^{-1}$  against the MSSA strain, while predicted to be the worst binder (Table 1). This may be related to structural rearrangements in the binding pocket, caused by the bulky side chain of  $\mathbf{7}$ , that mitigate the effect of RNA-methylation on the binding

_	Species	Strain description		ERY	CTY	TEL	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20
S. aureus		DSM20231 MIC	міс	0.5	0.5	0.3	4	0.5	1	0.1	0.1	8	4	4	32	1	4	4	8	8	0.3	4	8	>64	>256
m-positive	o. uurcus	DOMEDEOT	0	-	(>64)	(>128)	-	-	(>32)	(>256)	(>256)	(>128)	(>8)	-	-	(>32)	(>8)	(>8)	(>4)	-	(>64)	(>8)	-	-	-
	S. aureus	BAA976 msr(A)	міс	64	16	32	8	0.5	1	1	0.5	16	16	2	64	2	4	2	4	8	0.5	8	16	>64	>256
					(>2)	(>1)		(>64)	(>32)	(>32)	(>64)	(>2)	(>2)	-		(>16)	(>8)	(>16)	(>8)	- 5	(>64)	(>4)		1.	-
	S. aureus	BAA977 iMLS <sub>B</sub> M	MIC 2	256	8	4	8	0.5	2	1	0.5	32	16	2	64	1	4	2	4	4	0.5	8	16	>64	>256
				-	(>4)	(>8)	- 050	(>64)	(>16)	(>32)	(>64)	(>1)	(>2)	-	-	(>32)	(>8)	(>16)	(>8)	- 04	(>64)	(>4)		- 04	- 050
	S. aureus	ATCC43300	MIC 225	>256	>256	>128	>256	>64	>64	>128	128	32 (>1)	>128	256	>256	>128	>64	>64	>64	>64	>64	64	64	>64	>256
	1			0.5	0.5	0.3	2	0.3	1	0.1	0.1	2	2	1	8	1	2	2	8	8	0.5	2	4	16	128
Gram	B. subtilis	DSM402	MIC	0.5	(>64)	(>128)	-	(>128)	(>32)	(>256)	(>256)	(>16)	(>16)			(>32)	(>16)	(>16)	(>4)		(>64)	(>16)	7	-	120
F					>32	>32		>32	>32	>32	>32	>32	>32			>32	>32	>32	>32		>32	>32			>32
	E. coli	ATCC25922	MIC	-	-	-	-	-			-			-	-	-	-	-	-	-	-		-	-	-
	L	lpxC; MB4902			>32	>32		16	>32	32	32	>32	>32			32	16	>32	>32		16	>32			>32
	E. coli		MIC	-	-	-	-	(>2)	-	(>1)	(>1)			-	-	(>1)	(>2)	-		-	(>2)		-	-	_
		4-10-MD5747			8	8		2	2	2	2	32	2			2	2	2	2		2	16			32
	E. coli	toIC; MB5747	MIC -	-	(>4)	(>4)	-	(>16)	(>16)	(>16)	(>16)	(>1)	(>16)	-	-	(>16)	(>16)	(>16)	(>16)	-	(>16)	(>2)	-	-	(>1)
-negative	A. baumanii ATC	ATCC 19606	MIC -		>32	>32		>32	>32	>32	32	>32	>32			>32	>32	>32	>32		32	>32			>32
		AI CC 19606	MIC	-	-	-	-	-	-	-	(>1)	-	-	-	-	-	-	-	-	-	(>1)	-	-	-	-
	P. aeruginosa ATCC 2785	ATCC 27052	MIC		>32	>32		>32	>32	>32	>32	>32	>32			>32	>32	>32	>32		>32	>32			>32
		AI CC 2/655	WIIC	_	-	-	_	-	-	-	-	-	-	_	-	-	-	-	-	-	-	-	-	_	-
	P. aeruginosa PAO397; PAO1 d	DAC 307: DAC 1 d	MIC -	_	>32	>32	_	32	>32	32	32	>32	32	_		>32	>32	>32	>32	_	32	>32			>32
		FA0337, FA01 u		_	-	-	_	(>1)	-	(>1)	(>1)	-	(>1)	-	_	-	-	-	-	-	(>1)	-	-	_	-
Gram	K. pneumoniae	neumoniae ATCC700603	MIC .		>32	>32		>32	>32	>32	>32	>32	>32		_	>32	>32	>32	>32		>32	>32		_	>32
Ē	N. prieumomae Aro	7110010000	0		-	-		-	-	-	-	-	-			-	-	-	-		-	-			-
	C. albicans	ATCC 90028	міс		>32	>32	_	>32	>32	>32	>32	>32	>32		-	>32	>32	>32	>32	_	>32	>8		-	>32
*		7.1.00 00020																							
Yeast	C. neoformans	ATCC 208821; H99	МІС	-	>32	>32	-	>32	>32	>32	>32	>32	>32	-	-	>32	>32	>32	>32	-	>32	>32		-	>32
		HEK293; ATCC			-	-		-	-		-		-			-		-			-	-			-
į	H. sapiens	CRL1573	CC50	-	>32	>32	-	>32	>32	>32	>32	>32	>32	-	-	>32	>32	>32	>32	-	>32	>32	-	-	>32
Eucariotic																									
15	H. sapiens	Red blood cell	HC10	-	>32	>32	-	>32	>32	>32	>32	>32	>32	-	-	>32	>32	>32	>32	-	>32	>32	•	-	>32
		MIC color code	[µq/mL]	1	>;	32			16 - 3	32			4 - 8			1 -	2		(	0.1 - 0.	5	r	ot de	termir	ned
	CC <sub>50</sub> /HC <sub>10</sub> color code [µg/mL]							not determined																	
	CC <sub>50</sub>	/nc <sub>10</sub> color code	[µg/mL	J	>,	32		not	ueter	mined															

Fig. 3. Activity against a variety of Gram-negative strains and yeast and the Gram-positive bacteria B. subtilis and the high priority pathogen S. aureus.(32) Minimal inhibitory concentrations (MIC) are given for erythromycin (ERY), clarithromycin (CTY), telithromycin (TEL), and all candidates. The three best candidates from the computational stage (4-6) are highlighted in red. Additionally,  $CC_{50}(H$ . sapiens HEK293), and  $HC_{10}(H$ . sapiens red blood cells) values are provided. Selectivity indices ( $CC_{50}$  value divided by MIC) are shown in parentheses below the respective MIC value. The selectivity indices are color coded together with the respective MIC value to facilitate readability. Not determined values are indicated by a (-).

affinity.

Next, antibacterial MIC data was determined against several Gram-negative pathogens, including type strains from A. baumanii (DSM 30007, ATCC 19606), P. aeruginosa (DSM 50071, and ATCC 27853), E. coli (DSM 30083, ATCC 11775, and ATCC 25922), and K. pneumoniae (ATCC 700603, see Table 2 and Fig. 3). Those four species are the top-ranked bacteria from the WHO priority pathogens list for the development of new antibiotics.(32) Similar to established macrolide antibiotics, most compounds display no activity against the panel of Gram-negative bacteria. A notable exception is once more compound 6, showing activity against A. baumannii (MIC between 8 and  $32~\mu \mathrm{g\,m\,L^{-1}}$ ). To a lesser degree, compounds 11 (MIC of  $64~\mu \mathrm{g\,m\,L^{-1}}$ ) and 16 (MIC between 16 and  $32~\mu \mathrm{g\,m\,L^{-1}}$ ) are also active.

In addition, the compounds were tested against Gramnegative strains that harbour mutations which affect the drug permeability through the bacterial membrane, including E. coli  $\Delta lpxC$  (variation of the lipopolysaccharide structure) and  $\Delta tolC$  (without the major efflux pump), and P. aeruginosa  $\Delta mexAB/CD/EF/JKL/XY$  (without five of the major efflux pumps). Interestingly, all compounds regain significant activity (up to MIC values of  $2 \mu \text{g mL}^{-1}$ ) if the major drug efflux pump is eliminated in E. coli. This also includes compound 20, which is otherwise not even active against Gram-positive bacteria. Variation of the outer membrane by  $\Delta lpxC$  had a lesser effect, with only compounds 3, 5, 6, 11, 12, and 16 regaining activity to MIC values of  $16-32 \,\mu\mathrm{g}\,\mathrm{mL}^{-1}$ . A similarly small effect was displayed for the P. aeruginosa strain that lacks five of the major drug efflux pumps. Only five compounds regain activity to MIC values of  $32 \,\mu\mathrm{g}\,\mathrm{mL}^{-1}(\mathrm{Fig.}\ 3)$ .

A similar outcome can be achieved by the co-administration

of sub-MIC levels of colistin, an antibiotic which interacts with the outer membrane component Lipid A, thereby increasing the permeability of the membrane for other drugs. Colistin showed synergistic activity with each of the tested acylides 6, 11, and 16 against E. coli and A. baumanii, increasing the activity to MIC values of 0.25 to  $1 \mu \text{g mL}^{-1}$ , at colistin concentrations of 10% of its MIC value (Table 2). Like the efflux pump mutants, activity with colistin against P. aeruqinosa could not be reestablished to the same degree as for E. coli or A. baumanii, suggesting that P. aeruginosa still has effective efflux mechanisms. The activity against the effluxdeficient E.  $coli\ \Delta tolC$  mutant suggests that the main cause of the resistance in Gram-negative bacteria is the efficient elimination of the compounds via efflux pumps. In addition, the activities against E. coli  $\Delta tolC$  agree with the IC<sub>50</sub> values, with most MIC values around  $2 \mu \text{g mL}^{-1}$  and IC<sub>50</sub> values  $< 0.5 \,\mu\text{M}$ . Exceptions are the weaker binders 7, 17, and 20, which also display  $IC_{50}$  values above  $1 \mu M$ , and MIC values at  $16-32 \,\mu\mathrm{g}\,\mathrm{mL}^{-1}$ . This suggests that the activity against S. aureus, which exhibits a wider range of MIC values, might be governed by permeability as well.

### **Discussion and Conclusions**

Our rational design strategy allowed us to successfully reduce the library of 19 derivatives of clarithromycin first to ten and then to three candidates  $(\mathbf{6}, \mathbf{4}, \mathbf{5})$  by calculating solubilities, membrane permeabilities, and binding affinities at three different levels of accuracy (end point methods with implicit solvent, FEP with explicit solvent, and QM/MM calculations). In vitro experiments confirmed that molecules  $\mathbf{6}$  and  $\mathbf{4}$  exhibit the lowest IC<sub>50</sub> of the whole set. The heuristic prioritization process allows fast and early decisions based on several

Table 2. Activity against the three top-ranked bacteria from the WHO Priority pathogens list.(32) Minimum inhibitory concentrations (MIC) are given for molecules 6, 11, and 16 in combination with sub-inhibitory concentrations of the membrane-weakening antibiotic colistin sulfate (COS). Erythromycin, clarithromycin, telithromycin and all other acylides were ineffective.

	MIC $(\mu \text{mol } L^{-1})$								
	A. baumanii	P. aeruginosa	E. coli						
	DSM 30007	DSM 50071	DSM 30083						
COS	0.125	0.5	0.125						
6	8.0	> 128	128.0						
<b>6</b> + COS $(0.1 \times MIC)$	0.25	> 128	32.0						
$6 + COS (0.5 \times MIC)$	0.25	16.0	0.5						
11	64	> 128	64						
<b>11</b> + COS $(0.1 \times MIC)$	1	> 128	64						
11 + COS ( $0.5 \times \text{MIC}$ )	0.25	64	2						
16	16	> 64	> 64						
<b>16</b> + COS ( $0.1 \times MIC$ )	0.25	> 64	16						
<b>16</b> + COS ( $0.5 \times \text{MIC}$ )	0.125	32	0.25						

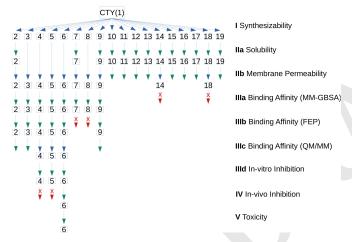


Fig. 4. Decision making process in the outlined rational design scheme. Each derivative of clarithromycin (CTY) must pass tests for synthesizability, solubility, membrane permeability, binding affinity at three different levels of accuracy (short MM-GBSA calculations, FEP simulations, and QM/MM calculations), *in-vitro* inhibition of the ribosome, *in-vivo* inhibition of resistant bacterial strains, and toxicity. Green arrows indicate molecules that meet the criteria and can be considered for further testing at the next level. Blue arrows indicate prioritization of molecules with especially beneficial properties. Crossed-out red arrows indicate molecules that have been eliminated because they significantly lack essential properties. The different levels of testing are ordered based on their (computational) expenses. The prioritization of molecules considers both the likelihood of exhibiting high activity and the available resources. Hypothetically, if all prioritized molecules fail to pass the test at one stage, the search should backtrack and continue with molecules with green arrows at a previous level until either a suitable drug candidate has been found, or all options have been exhausted.

approximations. In theory, only three molecules had to be synthesized and tested to find the best candidate (6). An overview of the decision making process is provided in Fig. 4.

Compounds **6**, **16**, and **11** exhibited some activity against the top three critical targets from the WHO priority pathogens list.(32) Further experiments demonstrated that *A. baumanii* and *E. coli* can be successfully inhibited by these compounds in combination with the membrane-weakening compound colistin,

or by elimination of the major efflux pump system. This confirms that the uptake through the outer membrane is a limiting factor in Gram-negative bacteria. In addition, the comparison with the membrane deficient *E. coli* and MSSA *S. aureus* strain suggests that, even in Gram-positive bacteria, permeability through the membrane affects the activity.

In vitro activity data for the high priority pathogen S. aureus allowed us to identify molecules 6, 3, and 16 as the most promising candidates with high activity, no cytotoxicity, and the ability to overcome existing macrolide resistance. Compared to telithromycin, lead compound 6 exhibits a 56-fold higher activity against S. aureus BAA 976, and an 8-fold increased activity against S. aureus BAA 977. Both activities are below the EUCAST clinical breakpoint of  $1 \mu \text{g mL}^{-1}$ , (35) which makes the molecule a good candidate to break the macrolide resistance in S. aureus. The comparison between the biophysical IC<sub>50</sub> data (Fig. 2) and cell-based MIC data (Fig. 3) indicates a good correlation for Gram-negative bacteria where the major efflux pump has been deactivated (E. coli  $\Delta tolC$ ). Nevertheless, several discrepancies can be observed for compounds with similar potencies in Gram-positive wild-type strains, like S. aureus. This suggests different uptake rates in Gram-positive bacteria. Compounds 6, 3, and 16 exhibit good MIC values.

The experimental data allow us to evaluate the design decisions, showing that an efficient strategy must account for more criteria than just binding affinity. Among the prioritized molecules in Table 1, the FEP and QM/MM calculations identified 80% of the active compounds in terms of IC<sub>50</sub> and 67% of the predictions were correct (SI Fig. S11). Strikingly, not a single molecule that was prioritized because of its high membrane permeability (2, 7, 14, 18, and, to a lesser degree 9) turned out to be effective. In contrast, almost all molecules that were selected for their hydrophilicity (3, 4, 5, 6, and, to a lesser degree, 8) were more active than clarithromycin. Thus, hydrophilicity turned out to be the better design criterion than high membrane permeability to reach low MIC values.

Some active compounds (in particular molecules 16 and to some degree also 11) were missed in the initial stages of the selection process. Molecule 16 almost progressed further, being among the six most hydrophilic molecules. However, MM-GBSA calculations predicted a low binding affinity for 16 (SI Table S3). Compound 11 was given a low priority because it ranked low both in terms of hydrophilicity and membrane permeability. It also would have exhibited a poor binding affinity in a posteriori MM-GBSA calculations (SI Table S3). Thus, it did not meet any of the considered requirements.

Any rational design scheme can only be as good as its design criteria. Future prioritization schemes could benefit from a better understanding of antibacterial activity against wild-type and efflux pump mutants, as well as other resistance mechanisms. A weighted scoring function that identifies the most important physico-chemical traits and their contributions to uptake and activity could guide the rational design process. Theoretical bounds with improved convergence properties could speed up the computational prioritization scheme. Additional in silico steps to evaluate the binding affinity of the compounds against different ribosome strains should be the next stage in the development of rational design strategies.

#### **Materials and Methods**

The simulations employed the CHARMM(36, 37) and NAMD(38) programs with the CHARMM36 force field.(39-41) The large ribosomal subunit (50S) from Deinococcus radiodurans (PDB codes: 2ZJR and 1J5A )(13, 14, 42) and all  $Mg^{2+}$  ions from the crystal structures were solvated in a cubic box with  $0.1\,\mathrm{M}$  KCl. The box size was equilibrated with constant pressure simulations and 20 ns of MD simulations were performed. The free energy simulations and QM/MM calculations are based on 5 ns simulations with Hamiltonian replica exchange (43, 44) of a truncated ribosome model. The free energy differences were analyzed with the FREN module of CHARMM.(45, 46) The implicit solvent calculations of solubility, (21, 30) membrane permeability, (25) and MM-GBSA binding affinity(27, 28) were conducted with the GBMV implicit solvent model. (20, 24) The QM/MM calculations were performed with the ChemShell package, (47) using the MNDO program (48) for the semi-empirical calculations with OM2-D3,(49-51) Turbomole(52) for the QM calculations with BP86-D3/Def2-SVP, (53, 54) and DL\_POLY(55) for the MM calculations.(56)

For the experimental testing, the compounds were synthesized as indicated in Fig. 1. The compounds were subjected to an  $IC_{50}$ determination assay against the E. coli ribosome. Cell lysates were prepared as described previously.(57) Luminescence measurements were carried out on a Labtech Clariostar microtiter plate reader. The stability of acylides in human blood plasma has been investigated as described before. (33) The determination of the minimal inhibitory concentrations and the cytotoxicity data was carried out as outlined in Refs. 58 and 59.

Detailed computational (Part 1) and experimental methodologies (Part 2, containing synthetic procedures, IC<sub>50</sub>-determination, MICdeterminations, cytotoxicity determination, KIT ComPlat IDs for most of the final compounds and further detailed information) are available as supplementary materials.

**ACKNOWLEDGMENTS.** This work was supported by the German Research Foundation (DFG) under Germany's Excellence Strategy - EXC 2033 - 390677874 - RESOLV. E. S.-G. also acknowledges the support of the Boehringer Ingelheim Foundation (Plus-3 Program) and of the collaborative research center CRC 1093 "Supramolecular Chemistry on Proteins", funded by the DFG. P. S. and E. S.-G. acknowledge the Gauss Centre for Supercomputing e.V. (www.gauss-centre.eu) for funding this project by providing computing time on the GCS Supercomputer JUQUEEN at Jülich Supercomputing Centre. F. S. was funded by the DFG Research Training Group GRK 2341 "Microbial Substrate Conversion" (Mi-Con). J.E.B. acknowledges funding from the Federal State of North Rhine-Westphalia and the European Regional Development Fund, for the Research Infrastructure "Center for System-based Antibiotic Research (CESAR)". A. Y. was supported by the European Research Council Grant 322581 "Novel Insights into Multi-Drug Resistance to Antibiotics and the Primordial Ribosome"; and the Kimmelman Center for Macromolecular Assemblies; A.Y. holds the Martin S. and Helen Kimmel Professorial Chair at the Weizmann Institute of Science.

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